

# Reliable optical characterization of e-beam evaporated TiO<sub>2</sub> films deposited at different substrate temperatures

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We studied e-beam evaporated TiO<sub>2</sub> films deposited at two different substrate temperatures between 120°C and 300°C. We reliably characterized the film samples on the basis of *in situ* and *ex situ* measurements. We carried out annealing on the samples and studied the induced changes in the properties of the films. The results can be useful for further laser-induced damage threshold investigations. © 2013 Optical Society of America

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## 1. Introduction

In recent years, titanium dioxide (TiO<sub>2</sub>) has been extensively studied because it is widely used as a high-index material in multilayer structures operating in visible and near infrared spectral ranges [1,2]. TiO<sub>2</sub> is used in combination with SiO<sub>2</sub> in optical coating production due to the high contrast of their refractive indices. TiO<sub>2</sub> exhibits interesting optical and electronic properties, chemical stability, good adhesion, and transparency in the visible and near infrared regions [3]. It is not surprising therefore, that many authors carry out research and publish results related to various aspects of TiO<sub>2</sub> amorphous films and crystalline structures. TiO<sub>2</sub> thin films can be produced by different deposition techniques including magnetron sputtering [4,5], RF sputtering [6,7], e-beam evaporation [8], reactive evaporation [3,9], plasma ion-assisted

deposition [10], e-beam evaporation with ion-beam assistance [11,12], plasmatron sputtering [13], electrophoretic deposition [14], and solgel methods [15]. Some authors compare the properties of TiO<sub>2</sub> films produced by different methods (see, for example, [16,17]). In many papers optical characterization of TiO<sub>2</sub> samples is performed [3,4,7,10,11,13,15,16,18]. Structural and morphological characterization is carried out by using various experimental techniques [5,7,10,11,13,14,16,17,19]. Influence of different deposition parameters on the optical, electrical, and structural properties can be found in [5,6,9,11–13,17]. Electronic properties of TiO<sub>2</sub>, including bandgap estimations, are obtained in the publications of several authors [7,8,12,14,15,20]. The optical and electronic properties of crystalline and amorphous TiO<sub>2</sub> structures calculated using quantum mechanical algorithms are reported in [19–21]. The effects of annealing on the optical, structural and electronic properties of TiO<sub>2</sub> films are considered by different researchers (see, for example, [6–8,10–12,17]).

The motivation for our research is based on the attractiveness of TiO<sub>2</sub> as high-index layer material in multilayer optical coatings for laser applications. The usage of TiO<sub>2</sub> in multilayers for lasers is restricted by its relatively low bandgap. It is known [22] that the bandgap values are decreasing with growing refractive index values, and for ultrashort pulses the laser-induced damage threshold (LIDT) of dielectrics is directly dependent on the bandgap of the high-index material [23–25] for sub-10-ps pulses [26]. TiO<sub>2</sub> films of relatively low packing density are supposed to have higher effective bandgaps than dense films and due to this they could be considered as high-index layers in coatings for laser applications. Therefore, an accurate study of the optical properties of e-beam evaporated TiO<sub>2</sub> films deposited under different deposition conditions would be useful for the potential applications of these films. In the present work we carefully characterized and compared the optical properties of e-beam evaporated TiO<sub>2</sub> films deposited at two different substrate temperatures. Also, we performed annealing of the TiO<sub>2</sub> samples and studied their effect on the optical constants and bandgap values of the TiO<sub>2</sub> films.

In Section 2 we describe experimental samples and measurement data. In Section 3 we provide a reliable optical characterization of the samples, which includes verification and validation of obtained results. In Section 4 we compare the optical constants and the bandgaps of the considered films before and after annealing. The final conclusions are presented in Section 5.

## 2. Experimental Samples and Measurement Data

For this study, four samples of single TiO<sub>2</sub> layers were prepared: two samples, named S1 and S3, on Glass B260 substrate of 1 mm thickness and two samples, named S2 and S4, on Suprasil substrate of 6.35 mm thickness. All four samples were produced by e-beam evaporation using SYRUSPro 710 deposition plant (Leybold Optics, Alzenau, Germany). Samples S1 and S2 were deposited at a relatively low substrate temperature of 120°. Samples S3 and S4 were prepared at a higher substrate temperature of 300°. The system was pumped down to 10<sup>-6</sup> mbar before deposition. Oxygen was fed near the Ti<sub>3</sub>O<sub>5</sub> targets to oxidize the evaporated material. The deposition rate was approximately 0.3 nm/s.

The SYRUSPro 710 deposition plant was equipped with broadband monitoring system (BBM) [27]. Samples S1 and S3 were placed on the calotte positions, where BBM measurements were performed. At the end of the depositions *in situ* transmittance data were recorded by the BBM device within a range from 400 to 950 nm with an approximately 0.5 nm wavelength step. The accuracy of *in situ* measurements  $\Delta T$  can be estimated as 1.0%.

Two weeks after the deposition, the transmittances and reflectances of all samples were measured *ex situ* using a Perkin Elmer 950 spectrophotometer. The measurement accuracy  $\Delta T$  is estimated as 0.2%. Measurements were performed in the spectral range from 320 to 1500 nm with a 1 nm wavelength step. Then, samples S2 and S4 were annealed at 300°C during 5 h. On the next day, the transmittance and reflectance spectra were measured again using a Perkin Elmer 950 spectrophotometer.

## 3. Characterization Results

In Fig. 1 we compare *in situ* and *ex situ* data related to samples S1 and S3. As evident from Fig. 1, the *ex situ* transmittance curves are shifted to the longer wavelengths with respect to those that are *in situ*. The shifts are more noticeable in the case of S1 sample deposited at low substrate temperature. These shifts can be explained by the structural difference between TiO<sub>2</sub> films in vacuum and in atmosphere. It is known that e-beam evaporated films have relatively low packing density [1]. After transferring these films from vacuum to atmosphere, their pores get filled with water. According to the Maxwell–Garnett formalism, this leads to an increase of the refractive index. As a consequence, the optical thickness of the film increases and the spectral curves shift to the longer wavelengths (a so-called vacuum shift) [1]. In the case of the S1 sample, this vacuum shift is larger than in the case of the S3 sample (Fig. 1).

We assume that the refractive indices and the extinction coefficients of the deposited films in the visible spectral range can be described by Cauchy and an exponential model, respectively [18]:

$$\begin{aligned} n(\lambda) &= A_0 + \frac{A_1}{\lambda^2} + \frac{A_2}{\lambda^4}, \\ k(\lambda) &= B_0 \exp[B_2\lambda^{-1} + B_3\lambda]. \end{aligned} \quad (1)$$

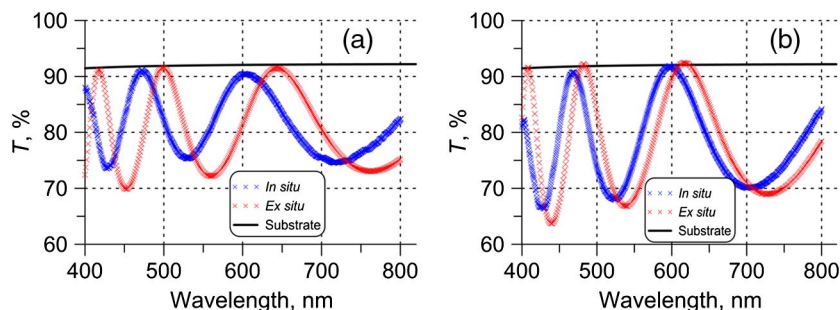


Fig. 1. Comparison of *in situ* and *ex situ* transmittance data related to samples S1 (a) and S2 (b).

The model parameters and film thickness  $d$  were found by minimization of a discrepancy function estimating closeness of experimental data to model spectral characteristics [18]. In the cases when only transmittance measurement data are available, only refractive index wavelength dependences and film thicknesses were determined [28]. The refractive index wavelength dependences of S1 and S3 TiO<sub>2</sub> films found from *in situ* and *ex situ* transmittance data are compared in Fig. 2. As expected, the refractive indices of the TiO<sub>2</sub> films in vacuum are lower than their refractive indices in atmosphere. The estimated values for physical thickness are quite close (see Table 1).

The offset  $\Delta n$  in a film refractive index caused by systematic errors in measurement transmittance data can be estimated as [29]

$$\Delta n = \frac{(n^2 + n_s)^3}{8nn_s(n_s - n^2)} \Delta T. \quad (2)$$

We use estimations of measurement accuracy presented in the end of Section 2. Substituting  $n = 2.21$  and  $n = 2.37$ ,  $n_s = 1.46$ , and  $\Delta T = 0.2\%$  in Eq. (2), we obtain that the accuracy of refractive index determination from *ex situ* data is about 0.6%. Substituting  $n = 2.1$  and  $n = 2.28$ ,  $n_s = 1.51$ , and  $\Delta T = 1.0\%$  into Eq. (2), we get accuracy of refractive index determination from *in situ* data that is about 3%.

To verify the obtained results, another characterization approach is applied. In the frame of this approach, the considered films are represented as a mixture of bulk material and voids. For any given wavelength, the film refractive index  $n$  can be described using Bruggemann's formula [30]:

$$n = \frac{1}{2} \sqrt{(3p - 1)n_b^2 + (2 - 3p)n_v^2 + \sqrt{((3p - 1)n_b^2 + (2 - 3p)n_v^2)^2 + 8n_b^2n_v^2}}, \quad (3)$$

where  $n_b$  and  $n_v$  are the refractive indices of the bulk material and voids, respectively;  $p$  is the volume fraction of the bulk material (packing density). In our

characterization process, we considered the refractive index of dense ion-beam sputtered (IBS) TiO<sub>2</sub> films [18] as bulk refractive index (see Fig. 2). In the case of *in situ* and *ex situ* data, voids are assumed to be filled with vacuum and water, respectively. In the course of the characterization process, only two unknown parameters,  $p$  and  $d$ , were searched for. The refractive indices determined from the *in situ* and the *ex situ* measurements related to samples S1 and S3 are depicted in Fig. 2 by dashed curves. The obtained values of  $p$  and  $d$  are presented in Table 1. It is evident from Fig. 2 and Table 1 that the parameters obtained by the two different models are in a good agreement, which is an indication of the reliability of our characterization results.

It is interesting that the packing densities found for films in vacuum and in atmosphere are quite close. This may be an indirect confirmation of the supposition that water fills the pores after exposing the samples to atmosphere. The film thicknesses in the case of S1 and S3 samples increased in atmosphere in comparison with vacuum by 0.9% and 0.8%, respectively. This may indicate "swelling" of the films under atmospheric conditions. In the last column of Table 1 the approximate values of film densities  $\rho$  are presented. These values were obtained by Lorentz-Lorenz formalism [3]:

$$\frac{n^2 - 1}{n^2 + 2} = C\rho, \quad (4)$$

where  $C = 0.1623 \text{ cm}^3/\text{g}$ . In Fig. 3 we show the theoretical dependence  $n(\rho)$  found from Eq. (4) and the positions of experimental values  $n(\rho)$  on this curve.

Characterization of the S2 and S4 samples is performed on the basis of transmittance and reflectance

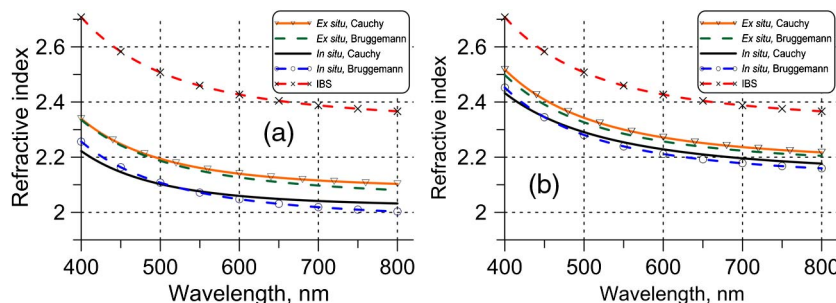


Fig. 2. Refractive indices of S1 (a) and S3 (b) films found from *ex situ* and *in situ* data using different models.

Table 1. Parameters of TiO<sub>2</sub> Films

Sample	Content	Model	Volume Fraction $p$ , %	Thickness $d$ , nm	$n$ at 500 nm	Density $\rho$ , g/cm <sup>3</sup>
S1	TiO <sub>2</sub> /Vacuum	Cauchy	74.1	453.9	2.06	3.2
		Bruggemann		444.7	2.1	
	TiO <sub>2</sub> /Water	Cauchy	73.5	457.9	2.18	3.43
		Bruggemann		456.5	2.19	
S2	TiO <sub>2</sub> /Water	Cauchy	74.9	457.3	2.21	3.48
		Bruggemann		458.7	2.20	
S3	TiO <sub>2</sub> /Vacuum	Cauchy	85.0	405.2	2.28	3.6
		Bruggemann		404.5	2.28	
	TiO <sub>2</sub> /Water	Cauchy	85.0	408.5	2.34	3.7
		Bruggemann		411.3	2.33	
S4	TiO <sub>2</sub> /Water	Cauchy	88.1	398.6	2.37	3.7
		Bruggemann		400.2	2.37	

index, but also the extinction coefficient has been reliably determined. The main parameters of the samples are collected in Table 1. It can be seen that the found refractive index values and the packing densities of S2 and S4 films are in good agreement with the refractive indices and the packing densities of S1 and S3 films, respectively. The slight differences in geometrical thicknesses can be attributed to different positions of S1, S3 and S2, S4 samples during the deposition process. The refractive indices and extinction coefficients of S2 and S4 films, found from *ex situ* data, are compared in Fig. 4. It can be seen that the refractive index of the S4 film is essentially higher than the refractive index of the S2 film produced at the relatively low substrate temperature. These observations are consistent with the results from [9,13]. The extinction coefficient of the S4 film is lower than the extinction coefficient of the S2 film. The densities of the films deposited at 300° (films S3 and S4) are higher than the densities of the TiO<sub>2</sub> films produced at 120° (films S1 and S2). These facts can be useful in the course of further study of potential applications of evaporated TiO<sub>2</sub> films.

#### 4. Influence of Thermal Annealing on the Properties of TiO<sub>2</sub> Thin Films

In Figs. 5 and 6, we compare the spectral characteristics of samples S2 and S4 before and after

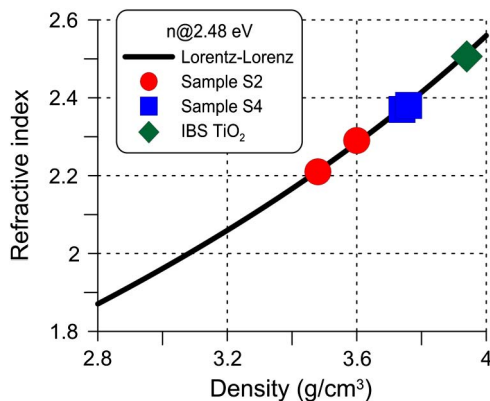


Fig. 3. Refractive indices of e-beam evaporated TiO<sub>2</sub> films at 500 nm before and after annealing (see Tables 1 and 2). The refractive index of IBS TiO<sub>2</sub> film is presented for comparison. Solid curve represents Lorentz–Lorenz dependence Eq. (4).

annealing. It can be seen that the transmittances of both samples decreased after annealing and the decrease is greater in the case when the sample was deposited at a lower substrate temperature. The reflectances of the samples after annealing exhibit oscillations of larger amplitudes than the reflectances of these samples before annealing. This effect is more pronounced in the case of sample S2. It can be seen from Figs. 5 and 6 that the spectral curves corresponding to the samples after annealing are shifted to longer wavelengths with respect to the spectral curves before annealing. Based on this qualitative analysis of the measured spectral characteristics, we expected that the refractive indices and the extinction coefficients of S2 and S4 films would increase after annealing.

In Fig. 7 we present the determined refractive indices and extinction coefficients of S2 and S4 films before and after annealing. As expected, the refractive indices and the extinction coefficients of both films increased after annealing. This increase is quite distinctive in case of sample S2, which was deposited at a relatively low substrate temperature. These results are consistent with the ones obtained in [10].

The main parameters of S2 and S4 films are collected in Table 2. One can see that the geometrical thicknesses of layer S2 after annealing have decreased by about 2.5%. That can be attributed to expulsion of water from the sample and subsequent changes of the film stoichiometry. For the same reason, the density and the packing density of film S2 increased after annealing. In the case of the S4 sample, the geometrical thickness, packing density, and density changed insignificantly, by 0.5%–1%, which is comparable with the characterization accuracy.

We estimated the bandgaps of the films with the help of Tauc plots [31–33]. In high-absorption spectral range, the absorption coefficient  $\alpha$  is related to the photon energy  $E$  in the following way [32,34,35]:

$$\alpha(E) = \frac{B(E - E_g)^m}{E}, \tag{5}$$

where  $B$  is a constant and  $m$  is a number characterizing the absorption process. This number takes the

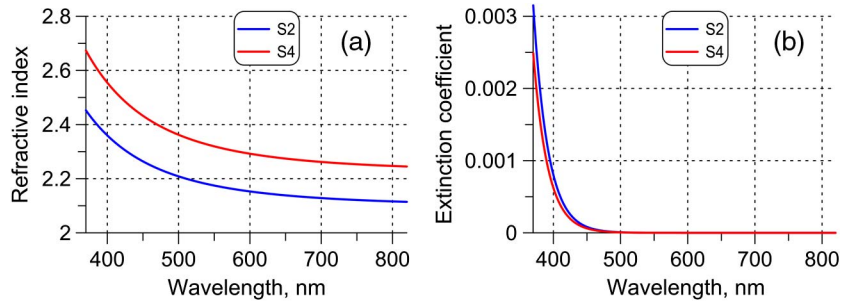


Fig. 4. (a) Refractive indices and (b) extinction coefficients of TiO<sub>2</sub> films deposited at different substrate temperatures.

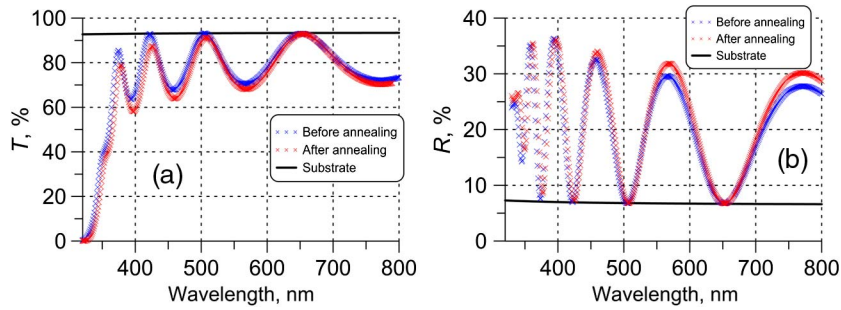


Fig. 5. (a) Transmittance and (b) reflectance spectra taken for sample S2 before and after annealing.

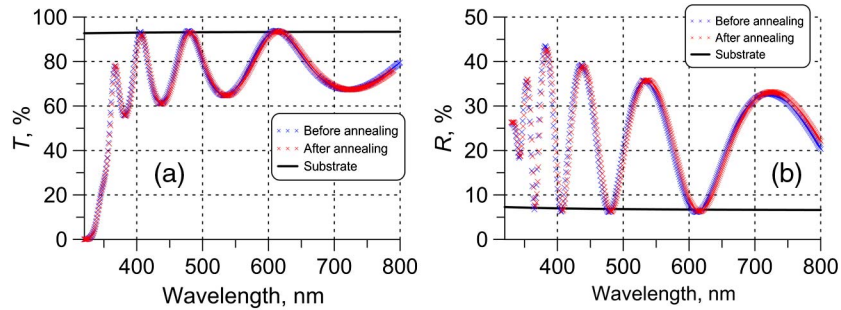


Fig. 6. (a) Transmittance and (b) reflectance spectra taken for sample S4 before and after annealing.

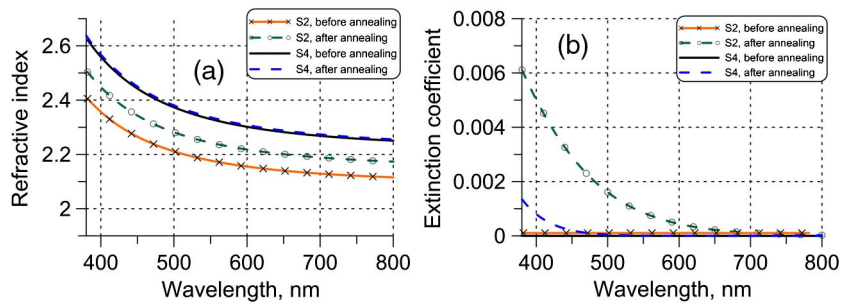


Fig. 7. (a) Refractive indices and (b) extinction coefficients of TiO<sub>2</sub> films before and after annealing.

Table 2. Parameters of TiO<sub>2</sub> Films before and after Annealing

Sample	$n$ at 500 nm		Thickness, nm		Density, g/cm <sup>3</sup>		Packing Density, %	
	Before	After	Before	After	Before	After	Before	After
S2	2.21	2.29	456.5	445.9	3.48	3.6	74.7	81.7
S4	2.37	2.38	398.6	400.9	3.7	3.75	88.1	89.0

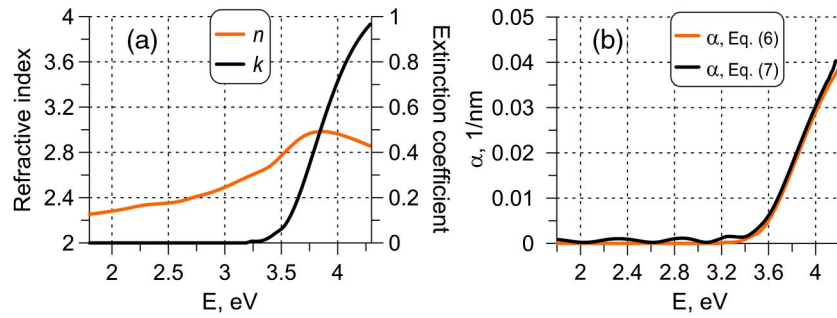


Fig. 8. (a) Model optical constants and (b) absorption coefficient of model sample calculated by Eqs. (6) and (7).

values 1/2 or 2 for direct and indirect transitions, respectively. If  $m = 1/2$  or  $m = 2$ , then  $(\alpha E)^2$  and  $(\alpha E)^{1/2}$  have linear dependences on  $E$ . Thus, by plotting  $(\alpha E)^2$  and  $(\alpha E)^{1/2}$  versus  $E$ , fitting them by linear dependencies and extrapolating this linear fit to zero, one obtains direct and indirect bandgap energies, respectively.

The absorption coefficient

$$\alpha = \frac{4\pi}{\lambda} k(\lambda), \quad E = \frac{1240}{\lambda}, \quad (6)$$

is expressed through extinction coefficient  $k(\lambda)$ . This coefficient is determined using a characterization process based on a thin film model. However, the characterization process in the short-wavelength spectral range requires the application of accurate models taking into account surface roughness. In addition to this, errors in  $T/R$  data affect the results. Hence, the dependence  $k(\lambda)$ , described by the exponential or the oscillator model and obtained from simultaneous fit of  $T/R$  data, might be considered as an estimation of extinction coefficient only. Another way to estimate absorption coefficient is to neglect the interference effects and scattering within the films and approximate  $\alpha$  on the basis of Beer's law.

To estimate the bandgaps on the basis of experimental transmittance data, first we estimated absorption coefficient on the basis of Beer's law. According to this law, the absorption coefficient may be approximated as [5,6,32,36]

$$\alpha \approx -\frac{1}{d} \ln(T). \quad (7)$$

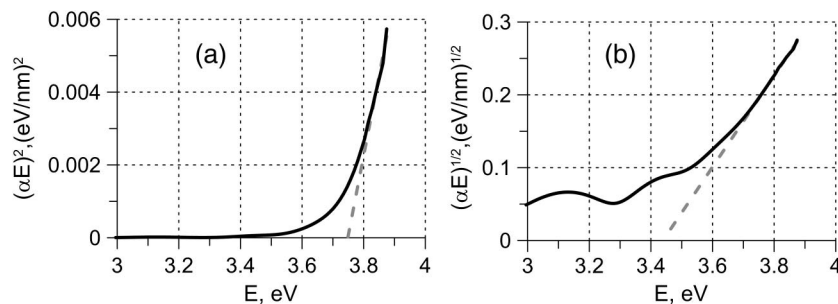


Fig. 9. Solid curves: (a) Tauc plots for the estimation of direct and (b) indirect band gaps; dashed lines: corresponding linear fits. The data are related to S2 sample after annealing.

To estimate the accuracy of the approximation in Eq. (7), we excluded the influence of measurement errors and used model data. Let us consider a model TiO<sub>2</sub> film of 400 nm thickness on a transparent substrate of 1 mm thickness. The refractive index  $n(\lambda)$  and the extinction coefficient  $k(\lambda)$  of this film in the spectral range from 1.8 to 4.3 eV are shown in Fig. 8(a). In Fig. 8(b), we compare the absorption coefficients calculated using Eqs. (6) and (7). It is evident that they are quite close to each other. The indirect bandgap values were found to be 3.33 and 3.29 eV, and direct bandgap values were found to be 3.78 and 3.81 eV. The deviations between  $E_g$  values calculated using Eqs. (6) and (7) do not exceed 0.05 eV. This value can be considered as an accuracy of bandgap determination in this study. The indirect bandgap is between 3.3 and 3.4 eV, and the direct bandgap is between 3.7 and 3.8 eV. The difference between direct and indirect  $E_g$  values is 0.3–0.4 eV, which is also in agreement with the results of other authors (see, for example [6,12,34]).

In this study we estimated both direct and indirect bandgaps. The  $E_g$  values found for sample S2 are 3.75–3.8 eV (direct) and 3.4–3.5 eV (indirect), and the  $E_g$  values found for sample S4 are 3.45–3.5 eV (indirect) and 3.7–3.75 eV (direct). In Fig. 9, the Tauc plots for direct and indirect bandgaps and the corresponding linear fits for the case of S2 sample after annealing are presented. According to [37], in amorphous TiO<sub>2</sub> films one should expect the indirect type of absorption coefficient behavior rather than the direct type. Thus, bandgap values calculated using Eq. (5) with  $m = 2$  should be considered as more reliable estimations. In our study we were interested

not only in the numerical values of the bandgaps, but also in the relative shift of the bandgap values after annealing of the samples. We found that for both samples, S2 and S4, the bandgap values decreased after annealing. The increase of the refractive indices and the respective decrease of the bandgap values are consistent with quantum mechanical calculations presented in [20].

The  $E_g$  values, estimated in the present study are in agreement with the results of other authors. In [12], the values of direct bandgap of TiO<sub>2</sub> films deposited by e-beam evaporation at 300°C are estimated as 3.81–3.92 eV. In [6], indirect and direct bandgap values are in the ranges 3.39–3.42 eV and 3.67–3.72 eV. In [8], the indirect and direct  $E_g$  values are 3.39–3.42 eV and 3.68–3.70 eV, respectively. In [34], the values 3.75 and 3.4 eV are presented for direct and indirect bandgaps, respectively. It should be noted also that the differences between direct and indirect  $E_g$  values obtained in the present study are about 0.3–0.4 eV, which is in agreement with the results from the literature (see, for example [6,8,12,34]). The estimation of the bandgaps shows that from the point of view of laser applications, the annealing of multilayer coatings containing TiO<sub>2</sub> layers does not benefit their optical resistance for ultrashort pulses.

## 5. Conclusions

The reliable characterization of e-beam evaporated TiO<sub>2</sub> films provides a basis for careful studying of LIDT in multilayer coatings containing TiO<sub>2</sub> layers. Also, predicting *in situ* refractive indices of TiO<sub>2</sub> films and the vacuum shift allows for more accurate monitoring of multilayer coatings with TiO<sub>2</sub> layers. Annealing of the films does not lead to an increase but to a decrease of the bandgap, which is not beneficial for obtaining high LIDT of optical coatings containing TiO<sub>2</sub> layers in sub-10-ps pulse regime. At the same time, the refractive index and packing density of TiO<sub>2</sub> films are increased, that may be useful for other applications, including operation with pulses having nanosecond range durations.

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